	SYLLABUS «Chemistry with the Foundations of Biogeochemistry» Degree of Higher Education – Bachelor Specialty <u>101 Ecology</u> Educational professional program - no Year of training – the forth; Semester: 7 Learning form – full-time Amount of the ECTC credits Language of instruction - English
Supervisor	Voitenko Larysa Vladyslavivna, Candidate of Chem Sci, Docent
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information (e-mail)	
eLearn Course URL	https://elearn.nubip.edu.ua/course/view.php?id=1105

DESCRIPTION OF COURSE

Discipline studies chemical, physical, geological and biological processes that are regulating the composition of the environment, biogeochemical cycles in their interaction with living matter through the biological systems of the Earth in time and space. The course includes the laws of the chemical composition formation of the ecosphere; principles of biogeochemical zoning, biogeochemical provinces and endemic diseases in them; theories of the origin of life, ways and types of biogenic and anthropogenic migration of chemical elements; methods for predicting chemical transformations of pollutants; mechanisms of isotope fractionation with living matter; the role of living matter in the geochemical processes of hypergenesis and crust weathering; biogeochemical patterns based on methods of chemical indication of the environmental state; transformation of xenobiotic.

Chapter	Hours (lectures/labs)	Results of learning	Tasks	Grading, scores	
	The fall semester, the 4 th year of study				
Module th	Module the 1 st . Biogeochemical characteristics of the ecosphere composition				
Chapter 1. Introduction. The object of research and the goal of the discipline. Life origin on the Earth: hypotheses and experimental.	4/6	 Know the subject and objectives of the course; areas of environmental issues related to biogeochemistry; modern ideas about the biochemical aspects of the life origin on the Earth (Oparin-Holden theory, RNA theory; progress in creating "artificial" life forms; safety in the chemical laboratory; Be able to work and analyze scientific and educational literature on the subject; Gain practical skills and be able to apply methods and equipment of rapid analysis on the example of determining the content of active chlorine. pH in drinking water; statistical processing of experimental results. 	Lab works processing and its defending; online testing on Elern platform	7	

STRUCTURE OF COURSE

ecosphere, the chemical elements andformulation of the biosphere concept in terms of its chemical structure and laws of function; Understand the role of living matterprocessi and defending	works 7
chemical elements and in terms of its chemical structure and laws of function; Understand the role of living matter	WOIKS /
chemical elementsin terms of its chemical structure and laws of function;andUnderstandthe role of living matterdefending	ing
elements and Understand the role of living matter defending	its
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biogeochemical as the main driving force of chemical	
laws transformation of the biosphere; the	
reasons for the existence of different	
types of classification of bioactive	
elements; biogeochemical functions	
of living matter.	
Apply quantitative laws (eg, Redfield	
atomic ratio, Le Chatelier principle)	
for environmental forecasts;	
Gain practical skills to perform	
analyzes of natural water pollutants	
(total iron content, nitrates)	
	works 20
Biogeochemical biogeochemical zoning and processi	
biogeochemical chains according to	-
Kowalski: types and causes of typical and	its
endemic diseases Rowalski; types and causes of typical endemic diseases, especially in defending	ng;
Ukraine; Module	
Understand the consequences of the control t	
anomalous distribution of chemical	lest
elements in the hydrosphere and	
lithosphere on the functioning of	
living matter; methods of prevention	
and treatment of endemic diseases;	
Gain practical skills of analytical	
determination of the content of	
fluorides, calcium, and magnesium in	
natural waters, the abnormal	
distribution of which is the cause of	
endemic diseases.	
Total the 1 st 16/16	34
module	54
Module the 2 ^d : Biogeochemical cycles of the main bioactive chemi	cal alamants
	works 16
founded of biogeochemical cycling:	0
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Biogeochemical the energy sources for the realization defending defending the biogeochemical cycling: the defending	ng:
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of sediment and elements cycling in ecosphere; how on	Elern
gaseous to apply chemical processes for the platform	1
biogeochemical organization of biogeochemical prevention of	
barriers doi the prevention of	
migration of ponatants (acta custe,	
barriers. Understand the experimental proofs	
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of alive matter role in cycling of	
chemical elements; degree of	
chemical elements; degree of anthropogenic pressure into	1
chemical elements; degree of anthropogenic pressure into biogeochemical cycles; why different	
chemical elements; degree of anthropogenic pressure into biogeochemical cycles; why different scientists give the various	
chemical elements; degree of anthropogenic pressure into biogeochemical cycles; why different scientists give the various quantitative assessment of pools,	
chemical elements; degree of anthropogenic pressure into biogeochemical cycles; why different scientists give the various quantitative assessment of pools, fluxes, exchange and reserve funds;	
chemical elements; degree of anthropogenic pressure into biogeochemical cycles; why different scientists give the various quantitative assessment of pools, fluxes, exchange and reserve funds; Gain practical skills of analytical	
chemical elements; degree of anthropogenic pressure into biogeochemical cycles; why different scientists give the various quantitative assessment of pools, fluxes, exchange and reserve funds; Gain practical skills of analytical determination of caffeine content in	
chemical elements; degree of anthropogenic pressure into biogeochemical cycles; why different scientists give the various quantitative assessment of pools, fluxes, exchange and reserve funds; Gain practical skills of analytical	

Chapter Chemistry preservatives xenobiotics2.	2/4	determination of active oxygen content and phosphates (as anthropogenic pollutants of natural waters) in laundry detergents and bleaching agents. Know the theoretical foundations of application and chemical mechanism of natural and artificial preservatives in food, cosmetic, pharmaceutical, wood-processing and other branches; their classification; risks of xenobiotic preservatives application; Understand the risks for human health of preservatives using; but why modern foods, cosmetic goods, medicine drugs etc., are impossible to safe without preservatives; Apply the knowledge of preservative safety and risks in everyday life; Gain practical skills of analytical determination of preservative E220 content (Sulfur dioxide) in foods and grape-contenting drinks.	Lab works processing and its defending; Module control test	20
Total the 2d module	14/14			36
All				70
Final testing				30
Finally				100

EVALUATION POLICY

Deadline policy and exam retake allowing:	Works that are submitted in violation of deadlines without good reason are evaluated at a lower grade. Retake of tests takes place with the lecturer's permission if there are good reasons (for example, student was sick and has the hospital sheet; took part in university event, scientific meeting etc.).	
Academic Integrity	Write-offs during tests and exams are prohibited (including using	
Policy:	mobile devices).	
Attendance Policy: Attendance is a mandatory component of the grade for which p are earned. For objective reasons (such international intern sickness), teaching may be provided on-line, in agreement with		
	Dean.	

GRADING SYSTEM

Rating of Higher	National grade according to the results of written examination		
education applicant,	exam	test	
scores			
90-100	Excellent	Pass	
74-89	Good		
60-73	Satisfactory		
0-59	Unsatisfactory	Fail	